

Letter to the Editor

Clearance and infusion

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Question: what must be the rate of infusion (K_0) in compartment 1 of any n-compartment pharmacokinetic model to achieve a specific steady-state concentration (C)?

Answer: $K_0 = C \times Cl$ where Cl is the total clearance.

Proof: For an *n*-compartment pharmacokinetic model with first-order kinetics except for the input rate K_0 which is constant and where elimination can take place from all compartments: let $x_i(t)$ be the amount of drug at time t in compartment i (i = 1,...,n).

With

$$K_j = \sum_{i=0}^{n} \sum_{j=i}^{n} k_{ji}$$
 $(j=1,...,n)$

$$\begin{cases} x'_{1} = -K_{1}x_{1} + k_{21}x_{2} + \dots + k_{n1}x_{n} + K_{0} \\ x'_{2} = k_{12}x_{1} - K_{2}x_{2} + \dots + k_{n2}x_{n} \\ \vdots \\ x'_{n} = k_{1n}x_{1} + k_{2n}x_{2} + \dots - K_{n}x_{n} \end{cases}$$
(1)

with $x_i(0) = 0$ for all i = 1, ..., n. Coefficient k_{ij} (in time⁻¹) stands for exchange from compartment i to compartment j (k_{i0} for outside).

The matrix M of this differential system is

$$M = \begin{pmatrix} -K_1 & k_{21} & k_{n1} \\ k_{12} & \neg -\overline{K}_2 - \cdots - \overline{k}_{n2} \\ \vdots & \vdots & \vdots \\ k_{1n} & k_{2n} & \cdots - K_n \\ \vdots & \vdots & \ddots & \vdots \\ \end{pmatrix} - - - \Delta$$

where Δ is the matrix obtained from M by discarding the first row and the first column of M.

Let $z_i(s) = L x_i(t)$, the Laplace transform of $x_i(t)$. Then Eq. (1) becomes

$$\begin{cases} sz_{1} = -K_{1}z_{1} + k_{21}z_{2} + \dots + k_{n1}z_{n} + (K_{o}/s) \\ \vdots \\ sz_{n} = k_{1n}z_{1} + \dots - K_{n}z_{n} \end{cases}$$
 (2)

Thus, by Cramer's method

$$z_{1}(s) = \frac{1}{P_{M}(s)} \begin{vmatrix} -K_{o}/s & k_{21} & \cdots & k_{n1} \\ 0 & -K_{2}-s & k_{n2} \\ \vdots & & & \\ 0 & k_{2n} & \cdots & -K_{n}-s \end{vmatrix}$$

where $P_M(s) = \det(M - sI_n)$ with I_n the unity matrix of order

Developing from the first column, we obtain

$$z_1 = -\frac{K_0}{sP_M(s)} \det(\triangle - sI_{n-1}) = \frac{P(s)}{O(s)}$$
 (3)

if
$$P(s) = -K_0 \det(\Delta - sI_{n-1})$$
 and $Q(s) = sP_M(s)$

From Heaviside's theorem if all eigenvalues of M are distinct

$$x_1(t) = \frac{P(o)}{Q'(o)} + \sum_i \frac{P(\lambda_i)}{Q'(\lambda_i)} e^{\lambda_i t}$$
(4)

where λ_i are the roots of $P_M(s)$ that is the eigenvalues of M, whose real parts are always negative, such as

$$\lim_{t\to+\infty} e^{\lambda_i t} = 0$$

Thus, at steady state

$$\lim_{t \to +\infty} x_1(t) = \frac{P(o)}{Q'(o)} = \frac{-K_0 \det \triangle}{P_M(o)} = \frac{-K_0 \det \triangle}{\det M}$$

(Note that if the eigenvalues of M are not distinct, the result for $\lim x_1(t)$ remains valid because Bromwich's theorem

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gives the same constant coefficient P(o)/Q'(o), while other entries include products of t and $e^{\lambda_t t}$ whose limit when $t \to +\infty$ is 0).

If V denotes the volume of distribution of compartment 1 and $C_1(t)$ the concentration in this compartment, then

$$\lim_{t \to +\infty} C_1(t) = \frac{-K_0 \det \triangle}{V \det M}$$
 (5)

For the concentration at steady state to have a specific value C, then Eq. (5) gives the value of the necessary infusion rate K_0

$$K_{o} = C \times \left(\frac{-V \det M}{\det \Delta}\right) \tag{6}$$

To obtain the complete solution to the problem, it remains to show that $-V \det M/\det \Delta$ represents the total clearance.

Since clearance does not depend on the route of administration in the model, we may obtain clearance by $Cl = D/(A \cup C_1)$ where $A \cup C_1$ is the area under concentration curve in compartment 1 after a bolus of a dose D in this compartment. Furthermore, $A \cup C_1 = Z_1(o)/V$ where $Z_1(s)$ denotes the Laplace transform of the amount $x_1(t)$ in compartment 1 receiving the bolus dose D

$$Z_{1}(s) = \frac{1}{P_{M}(s)} \begin{vmatrix} -D & k_{21} & \cdots & k_{n1} \\ 0 & -K_{2} - s & k_{n2} \\ 0 & k_{n2} & -K_{n} - s \end{vmatrix}$$

$$=\frac{-D\det\left(\triangle-sI_{n-1}\right)}{P_M(s)}$$

Thus

$$Z_1(o) = \frac{-D \det \triangle}{P_M(o)} = \frac{-D \det \triangle}{\det M} \text{ and } A \cup C_1 = \frac{-D \det \triangle}{V \det M}$$

so that

$$Cl = \frac{-V \det M}{\det \triangle}$$
 and $K_o = C \times Cl$

Note that the units are consistent on both sides of the equation (m = mass, l = liter, h = hour): K_0 is in m/h; C is a concentration in m/l; a clearance Cl is in l/h thus $C \times Cl$ is in $m/l \times l/h = m/h$ as K_0 .